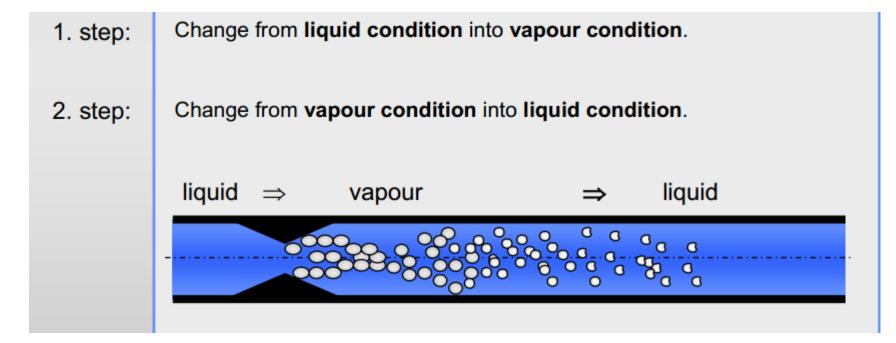
LECTURE 5: Cavitation in pipelines and in pumps, NPSH (net positive suction head)

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CAVITATION

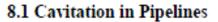
Cavitation is a physical process, which can arise in liquids. It describes the phase transition between the liquid and the vapour condition.

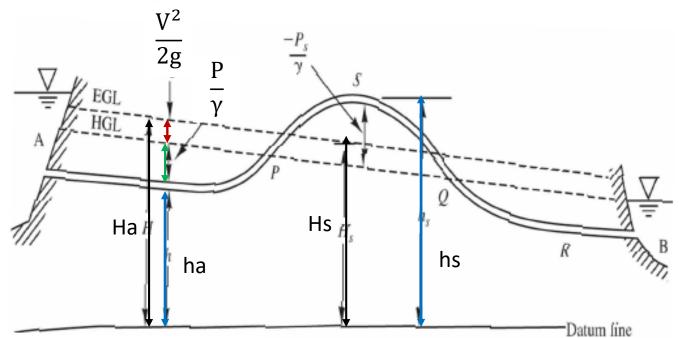
The process of cavitation has two steps:



CAVITATION IN PIPELINES

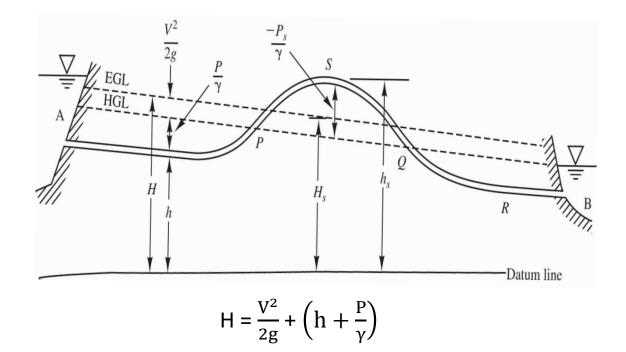
HA= Hs+ hL <---- Energy equation





h= Pipe axis level at point A

H= Total head at point A



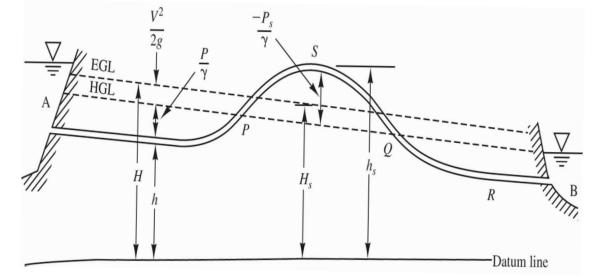
Total Head for point A \rightarrow Ha = $\frac{Va^2}{2g} + \left(ha + \frac{Pa}{\gamma}\right)$ Total Head for point S \rightarrow Hs = $\frac{Vs^2}{2g} + \left(hs + \frac{Ps}{\gamma}\right)$

Ha=HS

 $\frac{V^2}{2g}$ and h_s is a fixed positive value and the summation of $h_s + \frac{V^2}{2g} > Hs$

Hence, pressure head can only be negative value.

$$\left(\frac{-P}{\gamma}\right) + h_s + \frac{V^2}{2g} = H_s$$
 = Total Head at summit point



NEGATIVE PRESSURE EXIST IN THE PIPELINE WHENEVER THE PIPELINE RAISED ABOVE THE HYDRAULIC GRADIENT LINE"

• Negative pressure reaches a maximum value at the summit , "



• Water flow from S to R must flow against pressure gradient.

Higher pressure at $R \rightarrow$ Lower pressure at S

Water always flows toward lower-energy locations.

CAVITATION IN PIPELINES

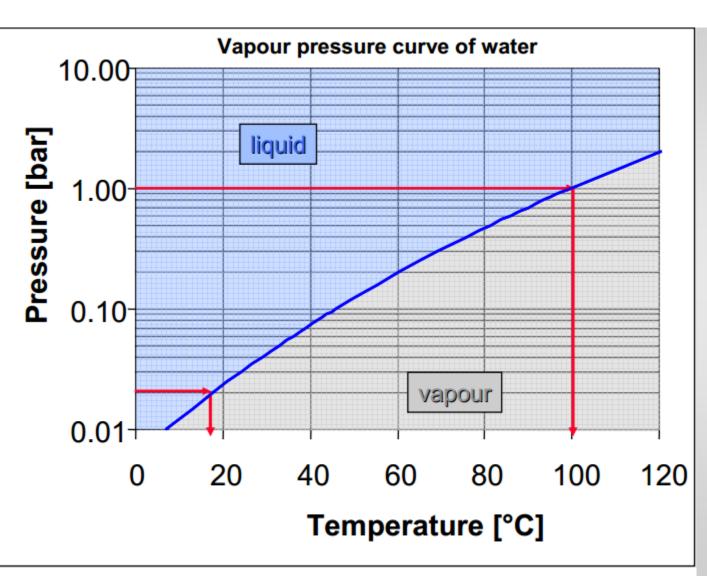
 If elevation head decrease > pressure head increase between S and R → water flows from S to R

$$Dh_{1-2} = \left(\frac{P_2}{\gamma} - \frac{P_1}{\gamma}\right) + h_{f12}$$
Elevation head decrease
Pressure head head loss between 1 to 2

It is important to maintain pressure at all points in a pipeline above the vapor pressure of water.

- Vapor pressure of water ≅ a negative water column of 10 m at 20⁰C.
- When the pressure in a pipe drops below vapor pressure of water
 - water will be vaporized
 - vapor pockets (cavitation) forms
 - separates water in the pipes
- These water pockets collapse in regions of higher pressure downstream.
- The action of vapor collapse is very violent, causing vibrations and sound that can greatly damage the pipeline.

Water Evaporation

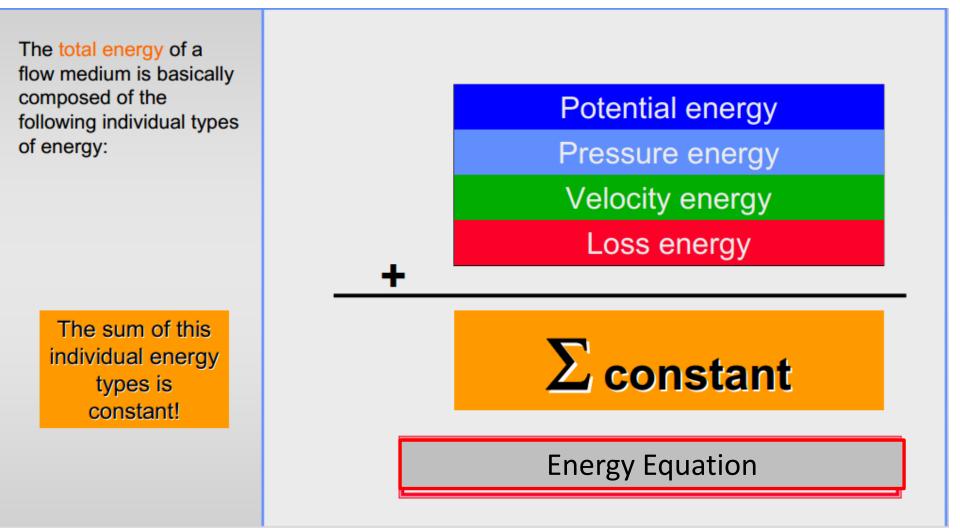


At atmospheric pressure (1 bar) water evaporates at 100°C.

When the pressure decreases, the evaporation process already starts at low temperatures.

Example:

At a pressure of 0.02 bar water evaporates already at a temperature of 18°C.

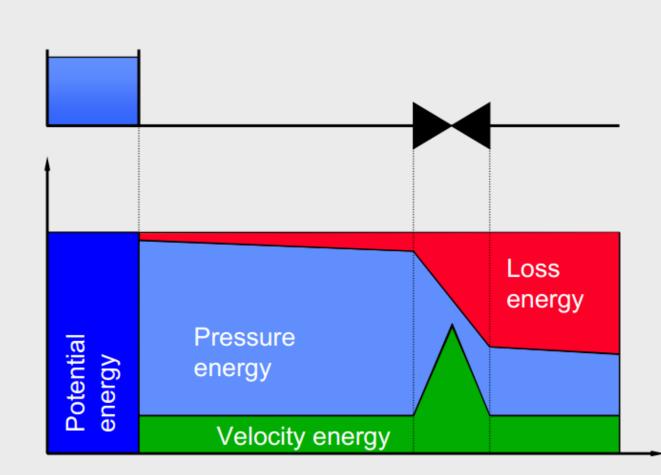


Energy forms in pipelines

In the store reservoir the existing total energy of the static flow is stored as potential energy.

In case of flow through a horizontal pipeline this available potential energy is converted into:

- velocity energy
- pressure energy
- loss energy

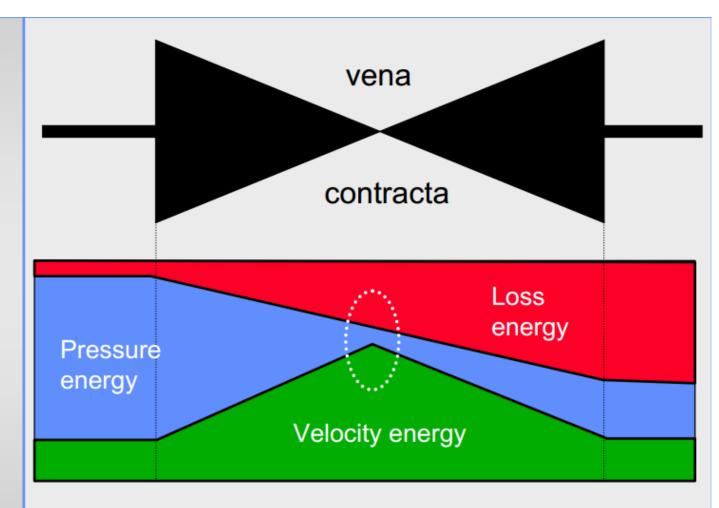


Change of energy forms at the throttling

Due to the contraction of the flow cross section at the throttling point, the flow velocity and thus the corresponding portion of energy rises considerably.

Due to throttling also the number of losses rises considerably.

At the vena contracta the remaining pressure energy and thus the local pressure decrease considerably because of the constancy of the total energy.



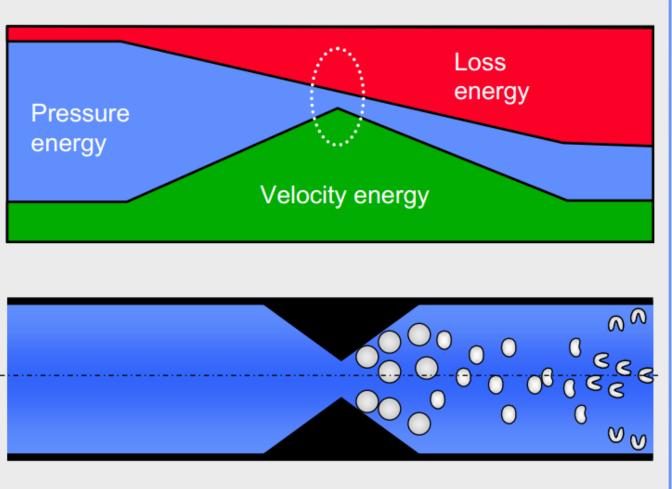
Change of energy forms at the throttling

If at this point the water pressure gets lower than the vapour pressure of the medium, it will evaporate.

There will be vapour bubbles, ...

... which are deformed under increasing pressure... ... and will finally

implode.





Typical cavitation damages

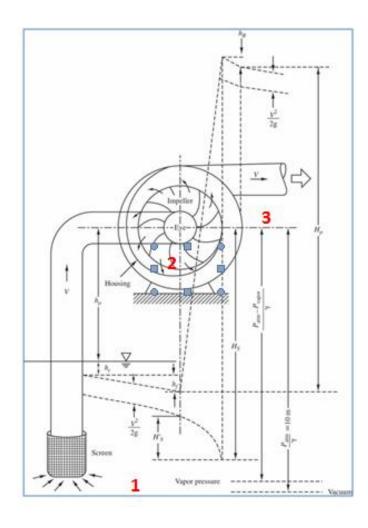
Cavitation damages at a butterfly valve.

Operating conditions:

- upstream pressure: 1.2 -1.4 bar
- downstream pressure: 0.1 bar
- flow velocity: 2.2 m/s (referred to DN)
- duration of operation: 2 years
- opening degree of disc: approx.: 30°



CAVITATION IN PUMPS



$$z_1 + \frac{V_1^2}{2g} + \frac{P_1}{\gamma} = z_2 + \frac{V_2^2}{2g} + \frac{P_2}{\gamma} + h_L$$
$$V_1 = 0, P_1 = P_{atm} V_2 \twoheadrightarrow V_s, P_2 \twoheadrightarrow P_S$$
$$h_p = z_2 - z_1$$

$$0 + \frac{P_{atm}}{\gamma} = \frac{V_{S}^{2}}{2g} + \frac{P_{S}}{\gamma} + h_{L(1-2)} + h_{P}$$

$$\frac{P_{S}}{\gamma} = \frac{P_{atm}}{\gamma} - \left(\frac{V_{S}^{2}}{2g} + h_{L(1-2)} + h_{P}\right)$$

NPSH (NET POSITIVE SUCTION HEAD)

Net Positive Suction Head available (NPSH_A) = $\frac{P_S}{\gamma} - \frac{P_V}{\gamma}$

$$= \frac{P_{atm}}{\gamma} - \frac{P_V}{\gamma} - \left(\frac{V_S^2}{2g} + h_{L(1-2)} + h_P\right)$$

Net Positive Suction Head Required (NPSH_R) =
$$\frac{P_S}{\gamma} - \frac{P_3}{\gamma} = H_S'$$

To avoid cavitation $P_3 > P_V \rightarrow NPSH_a > NPSH_R$

$$\frac{P_{S}}{\gamma} - \frac{P_{V}}{\gamma} > \frac{P_{S}}{\gamma} - \frac{P_{3}}{\gamma}$$

Safety factors other dissolved gases in water, $NPSH_a - 0.6 \text{ m} > NPSH_R$

NPSHavaliable **vs NPSH**required

- NPSH_{avaliable} is calculated for each system
- NPSH_{required} is given by pump manufacturer

The formula for calculating NPSHA:

$NPSH_A = H_A \pm H_Z - H_F + H_V - H_{VP}$

Term	Definition	Notes
H _A	The absolute pressure on the surface of the liquid in the supply tank	 Typically atmospheric pressure (vented supply tank), but can be different for closed tanks. Don't forget that altitude affects atmospheric pressure (H_A in Denver, CO will be lower than in Miami, FL). <u>Always</u> positive (may be low, but even vacuum vessels are at a positive <u>absolute</u> pressure)
Hz	The vertical distance between the surface of the liquid in the supply tank and the centerline of the pump	 Can be positive when liquid level is above the centerline of the pump (called static head) Can be negative when liquid level is below the centerline of the pump (called suction lift) Always be sure to use the lowest liquid level allowed in the tank.
H _F	Friction losses in the suction piping	 Piping and fittings act as a restriction, working against liquid as it flows towards the pump inlet.
H_V	Velocity head at the pump suction port	 Often not included as it's normally quite small.
H _{VP}	Absolute vapor pressure of the liquid at the pumping temperature	 Must be subtracted in the end to make sure that the inlet pressure stays above the vapor pressure. Remember, as temperature goes up, so does the vapor pressure.